Naming Ionic compounds containing transitional metals

Transitional	l metal	Lovidation	number in	roman	numerals	non-metal
Transitional	ımetai	loxidation	number in	roman	numerais	i non-metai

FeO = Iron(II)Oxide	FeO = I	lron(II)Oxide
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Why? Because oxygen is in group 6 and gains 2 electrons (-2). Therefore, those two electrons were lost by iron (+2)

Practice Naming these compounds?

CuCl

HgO

 PbO_2

 $Ti(NO_3)_4$